

Ionic Compounds Matching Game

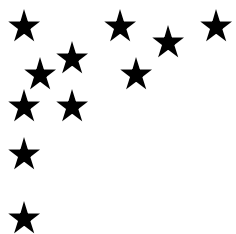
Ways to use Matching Game Cards:

- *Classroom skill-building games.*
- *Group activities.*
- *Individual, targeted practice.*
- *Study stations.*
- *Bell-ringer activity.*
- *Unit review.*
- *The possibilities are endless! You will come up with your own uses.*

Tips for making this product a resource you can use year after year:

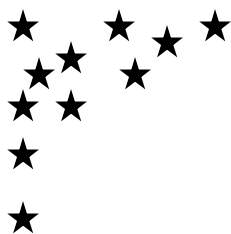
1. Print out all the pages containing the Matching Game Cards - there are six cards per page.
2. Cut out each individual card, along the borderlines.
4. Place the cards in a laminating folder. You can fit six cards per laminating folder.
5. Laminate the cards.
6. Cut out the laminated cards - you now have a set of laminated Matching Game Cards.

This set contains 48 Matching Game Card pairs.



Barium chloride

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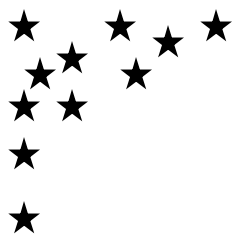
Magnesium oxide

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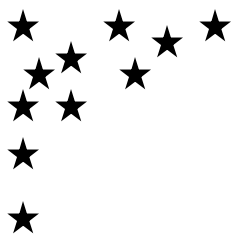
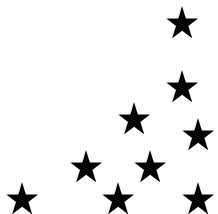
Lithium oxide

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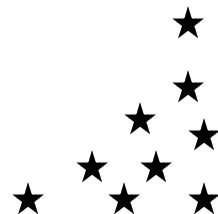
Calcium fluoride

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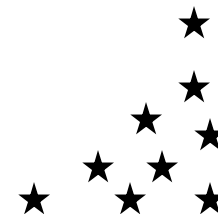
Aluminum bromide

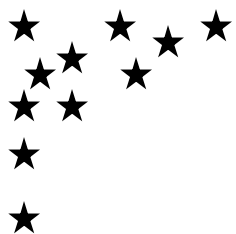
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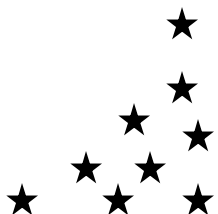
Aluminum sulfide

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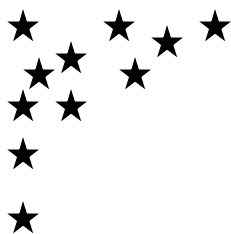




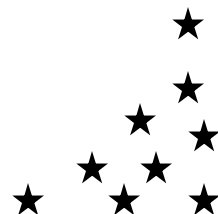
Potassium nitride



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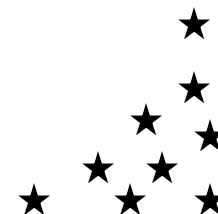
Potassium chloride



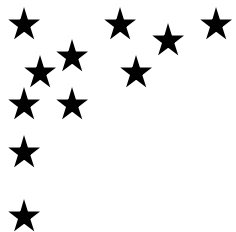
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Barium sulfate



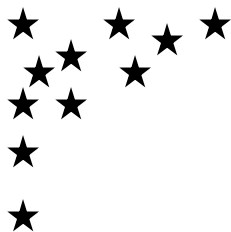
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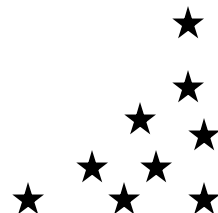
Magnesium bromide



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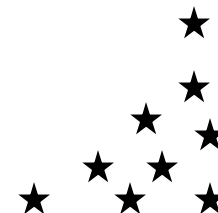
Lithium carbonate



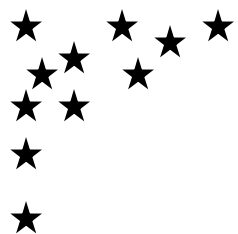
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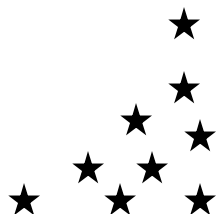
Potassium iodide



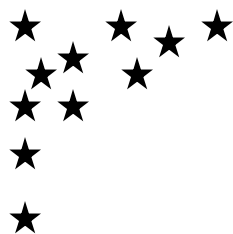
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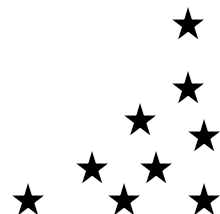
Copper (II) hydroxide



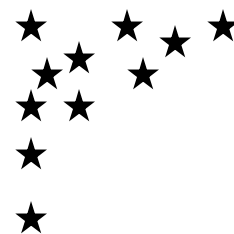
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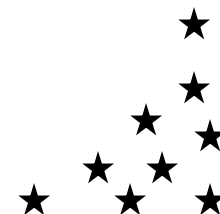
Ammonium chloride



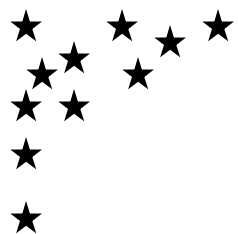
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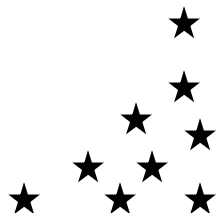
Iron (III) nitrate



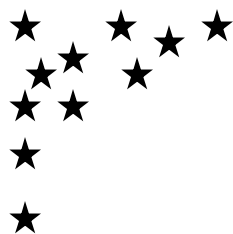
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Copper (I) phosphide



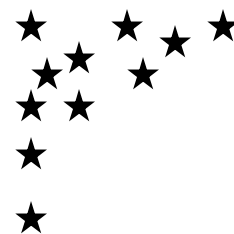
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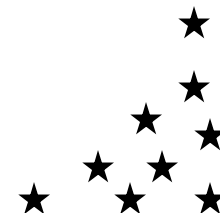
Tin (IV) selenide



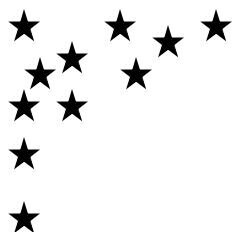
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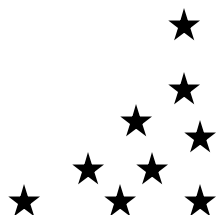
Lead (IV) sulfate



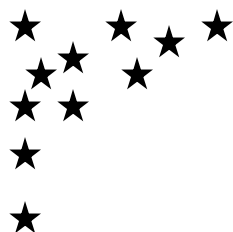
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Manganese (III) sulfite



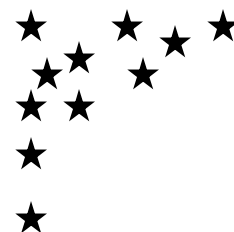
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Aluminum cyanide



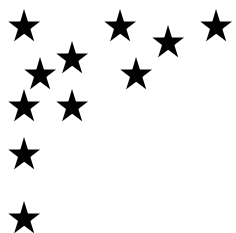
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Chromium (VI) phosphate



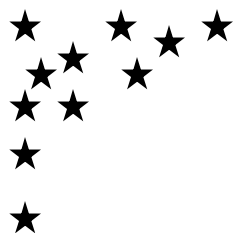
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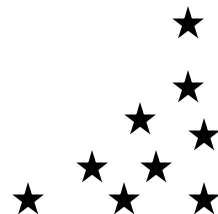
Tin (II) nitrite



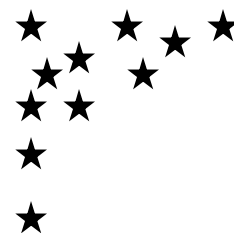
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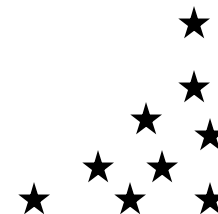
Cobalt (III) oxide



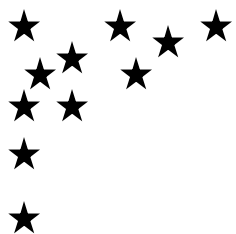
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Chromium (III) hydroxide

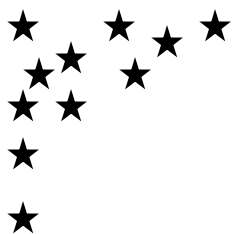
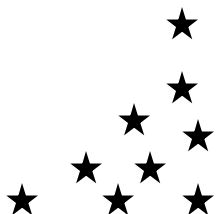


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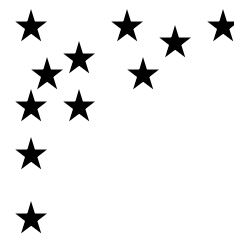
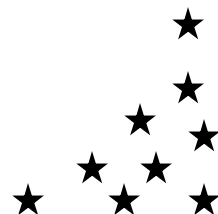
Lithium iodide

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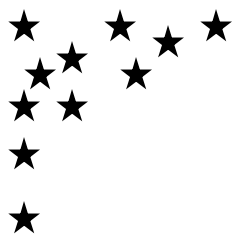
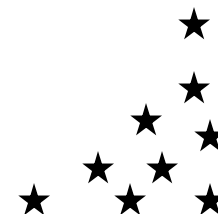
Lead (II) nitride

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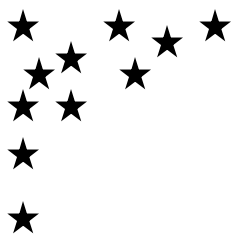
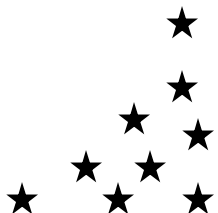
Silver bromide

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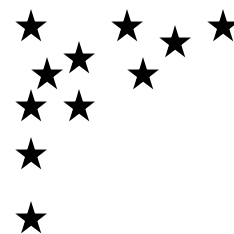
Aluminum oxide

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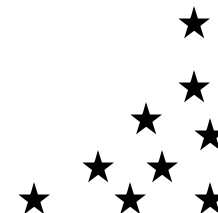
Titanium (II) cyanide

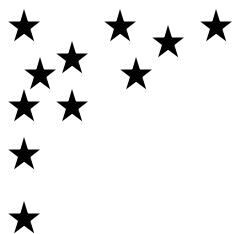
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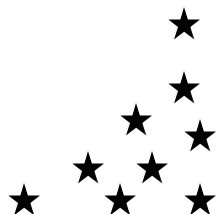
Strontium sulfate

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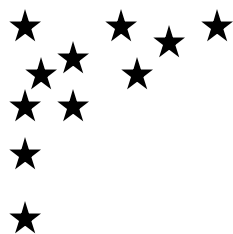




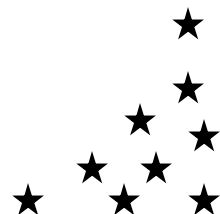
Barium chlorate



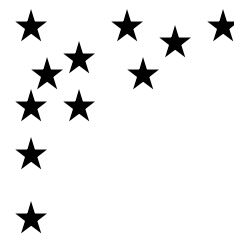
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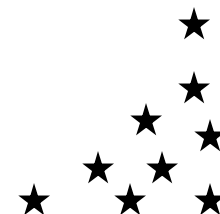
Magnesium hydroxide



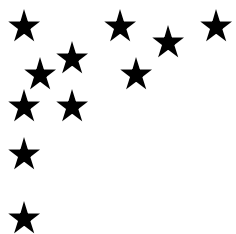
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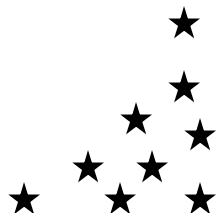
Mercury (I) oxide



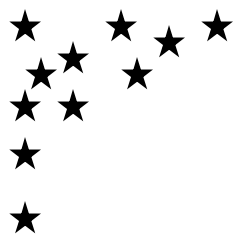
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Copper (I) oxide



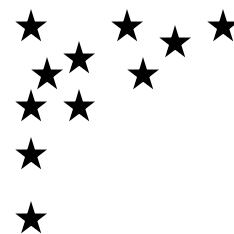
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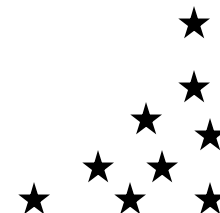
Sodium acetate



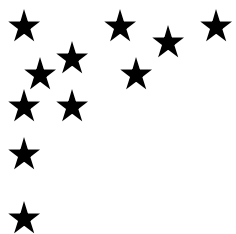
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Beryllium sulfide



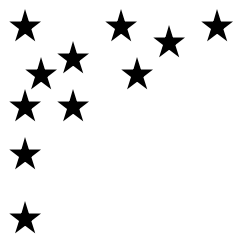
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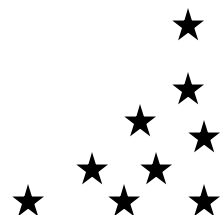
Sodium cyanide



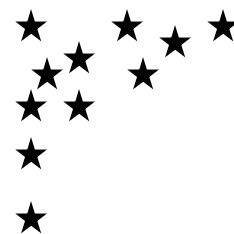
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Lead (IV) oxide



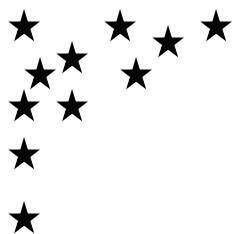
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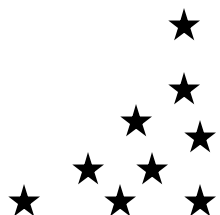
Sodium bromide



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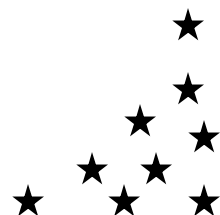
Gold (I) nitrate



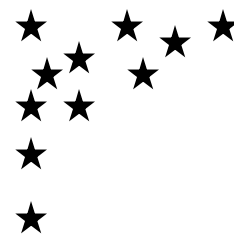
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Sodium chromate



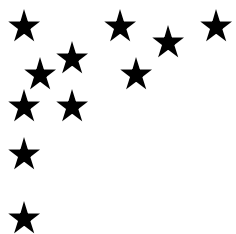
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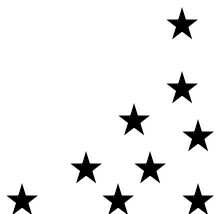
Barium nitrate



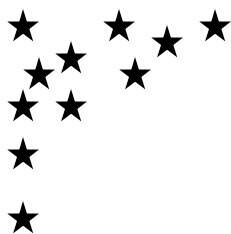
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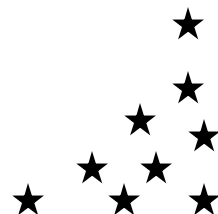
Ammonium sulfate



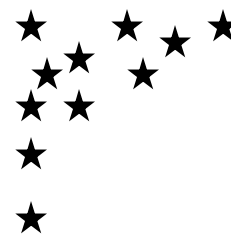
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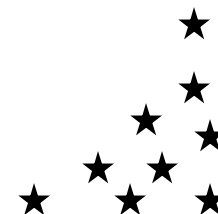
Aluminum hydroxide



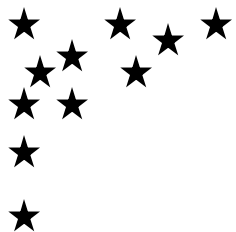
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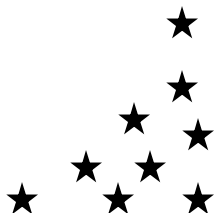
Calcium phosphate



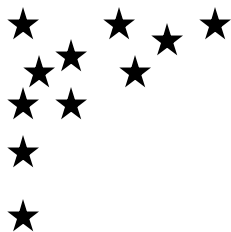
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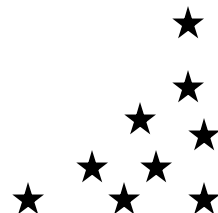
Cesium cyanide



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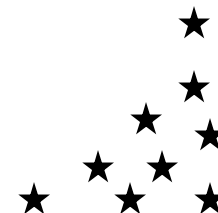
Sodium nitrate



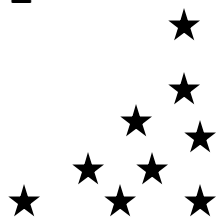
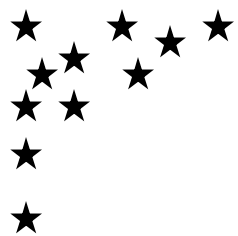
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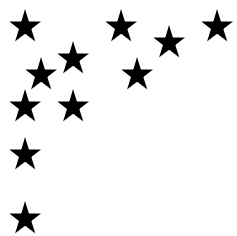
Calcium acetate



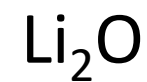
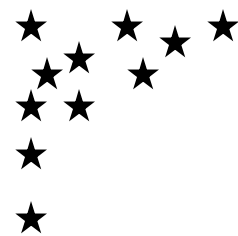
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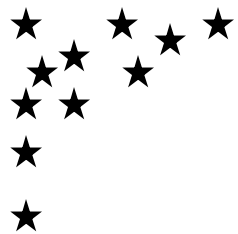
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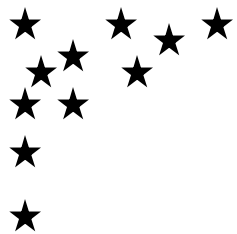
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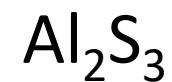
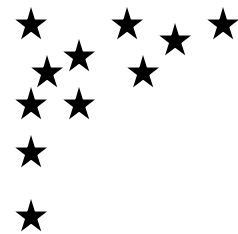
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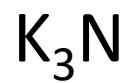
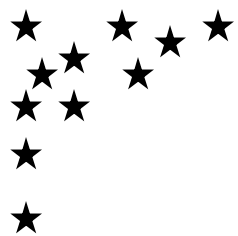
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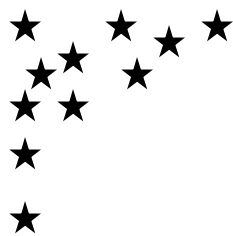
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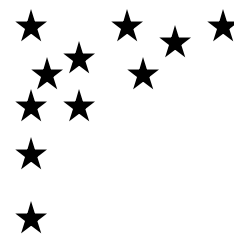
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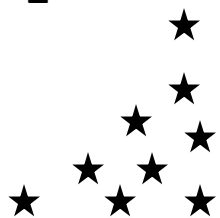
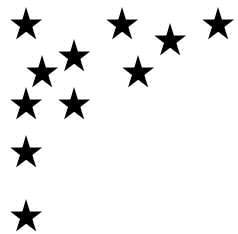
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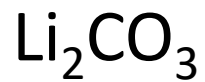
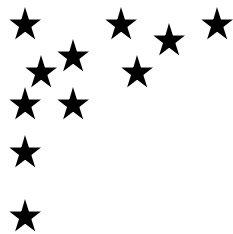
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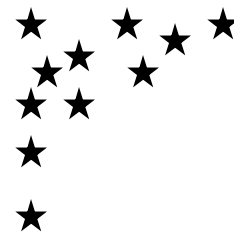
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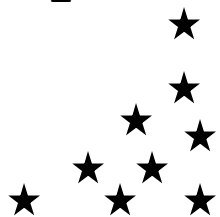
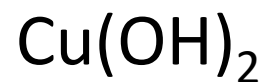
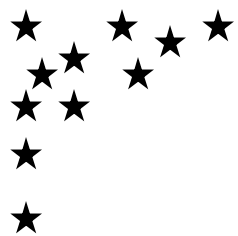
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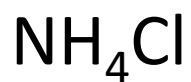
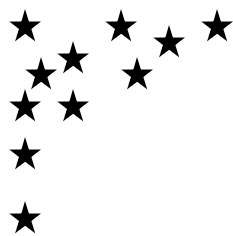
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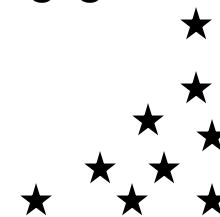
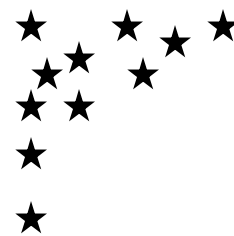
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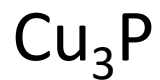
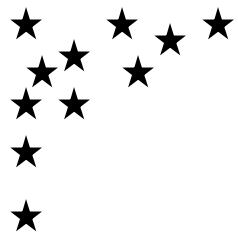
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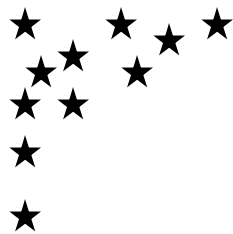
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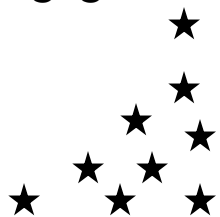
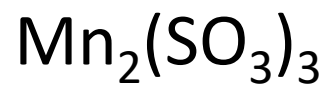
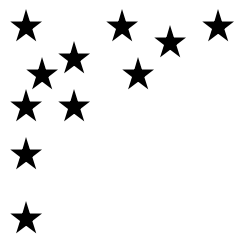
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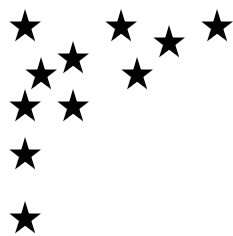
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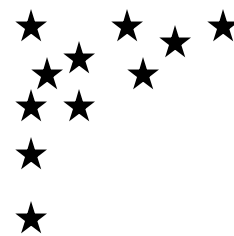
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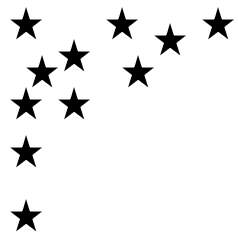
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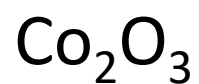
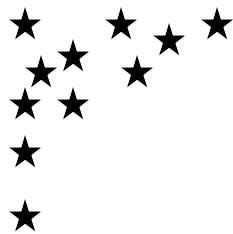
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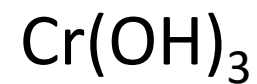
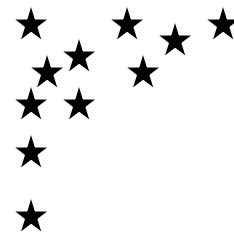
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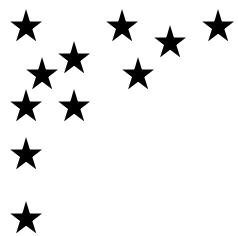
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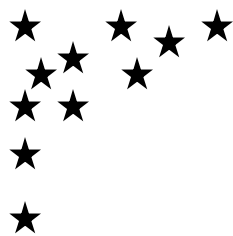


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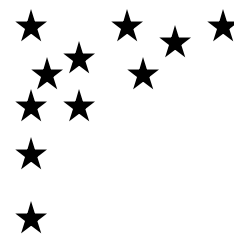
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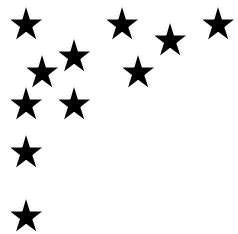
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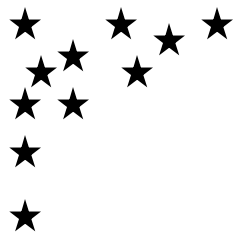
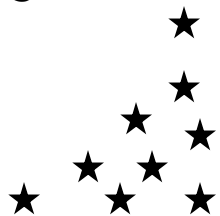
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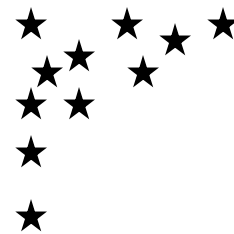
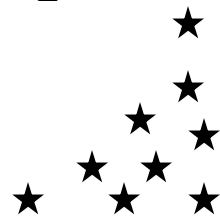
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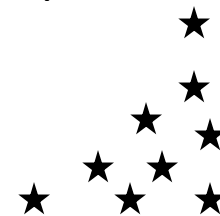
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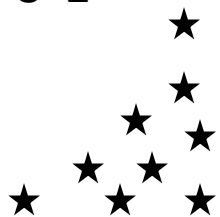
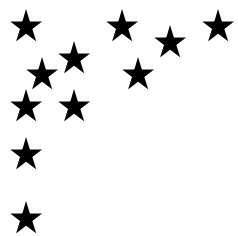
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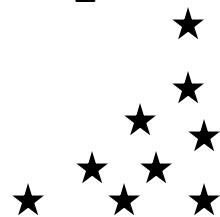
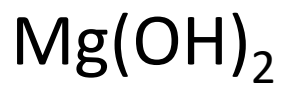
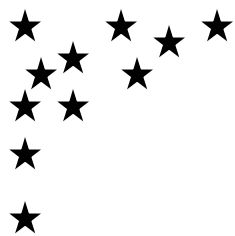
SrSO_4

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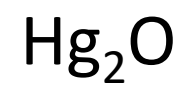
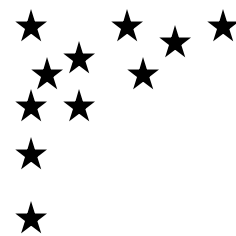




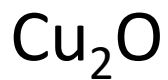
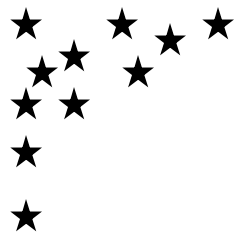
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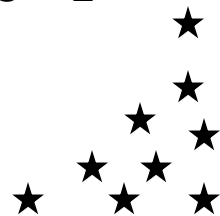
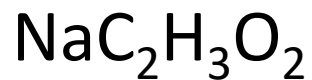
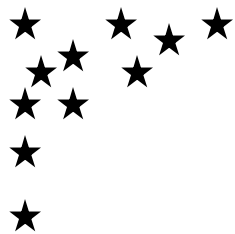
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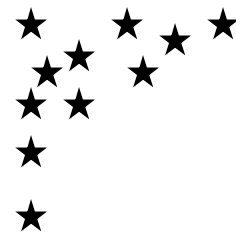
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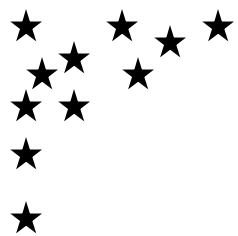
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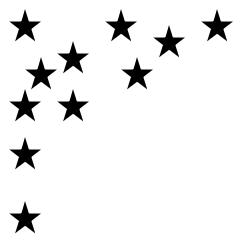
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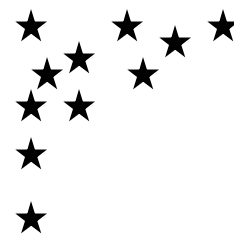
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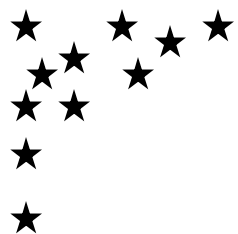
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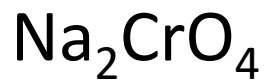
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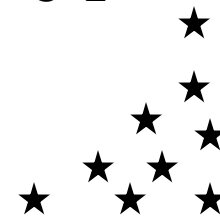
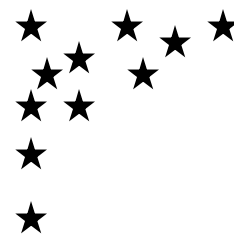
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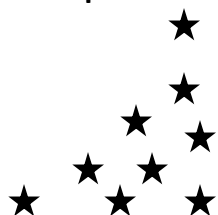
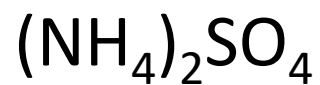
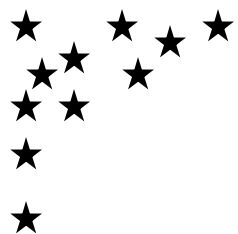
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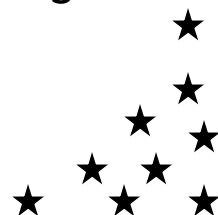
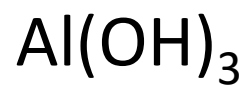
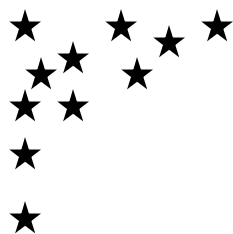
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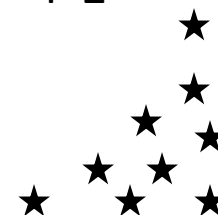
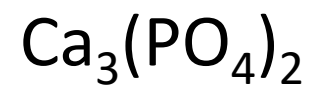
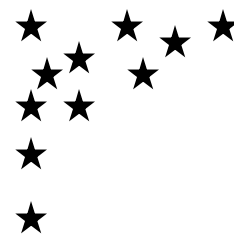
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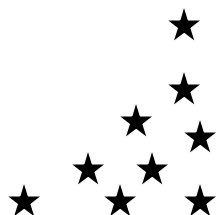
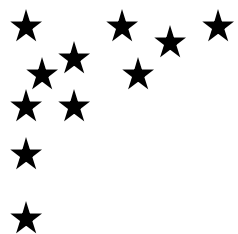
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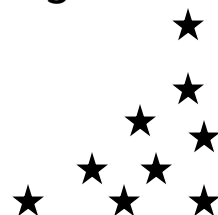
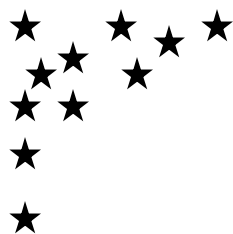
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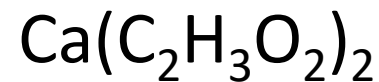
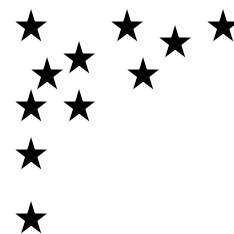
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Answer Key

1	Barium chloride	BaCl ₂
2	Magnesium oxide	MgO
3	Lithium oxide	Li ₂ O
4	Calcium fluoride	CaF ₂
5	Aluminum bromide	AlBr ₃
6	Aluminum sulfide	Al ₂ S ₃
7	Potassium nitride	K ₃ N
8	Potassium chloride	KCl
9	Barium sulfate	BaSO ₄
10	Magnesium bromide	MgBr ₂
11	Lithium carbonate	Li ₂ CO ₃
12	Potassium iodide	KI
13	Copper (II) hydroxide	Cu(OH) ₂
14	Ammonium chloride	NH ₄ Cl
15	Iron (III) nitrate	Fe(NO ₃) ₃
16	Copper (I) phosphide	Cu ₃ P
17	Tin (IV) selenide	SnSe ₂
18	Lead (IV) sulfate	Pb(SO ₄) ₂
19	Manganese (III) sulfite	Mn ₂ (SO ₃) ₃
20	Aluminum cyanide	Al(CN) ₃
21	Chromium (VI) phosphate	Cr(PO ₄) ₂
22	Tin (II) nitrite	Sn(NO ₂) ₂
23	Cobalt (III) oxide	Co ₂ O ₃
24	Chromium (III) hydroxide	Cr(OH) ₃

25	Lithium iodide	LiI
26	Lead (II) nitride	Pb ₃ N ₂
27	Silver bromide	AgBr
28	Aluminum oxide	Al ₂ O ₃
29	Titanium (II) cyanide	Ti(CN) ₂
30	Strontium sulfate	SrSO ₄
31	Barium chlorate	Ba(ClO ₃) ₂
32	Magnesium hydroxide	Mg(OH) ₂
33	Mercury (I) oxide	Hg ₂ O
34	Copper (I) oxide	Cu ₂ O
35	Sodium acetate	NaC ₂ H ₃ O ₂
36	Beryllium sulfide	BeS
37	Sodium cyanide	NaCN
38	Lead (IV) oxide	PbO ₂
39	Sodium bromide	NaBr
40	Gold (I) nitrate	AuNO ₃
41	Sodium chromate	Na ₂ CrO ₄
42	Barium nitrate	Ba(NO ₃) ₂
43	Ammonium sulfate	(NH ₄) ₂ SO ₄
44	Aluminum hydroxide	Al(OH) ₃
45	Calcium phosphate	Ca ₃ (PO ₄) ₂
46	Cesium cyanide	CsCN
47	Sodium nitrate	NaNO ₃
48	Calcium acetate	Ca(C ₂ H ₃ O ₂) ₂

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